

### 3. Sınav için denklemler ve sabitler

$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$	$pK_a = -\log [K_a]$
$R = 8,315 \text{ J K}^{-1} \text{ mol}^{-1}$	$pOH = -\log [OH^-]$
$\mathfrak{F}$ (Faraday sabiti) = $96,485 \text{ C mol}^{-1}$	$pH = -\log[H_3O^+]$
$1V = 1 \text{ J/C}$ $1A = 1C/s$	$pH \cong pK_a - \log\left(\frac{HA}{A^-}\right)$
$K_{su} = 1,00 \times 10^{-14}$ , $25^\circ\text{C}$ de	$\Delta E^\circ(\text{pil}) = E^\circ(\text{katot}) - E^\circ(\text{anot})$
$14,00 = pH + pOH$ , $25^\circ\text{C}$ de	$RT/\mathfrak{F} = 0,025693 \text{ V}$ , $25,00^\circ\text{C}$ de
$\Delta G^\circ = -RT \ln K$	$\mathfrak{F}/RT = 38,921 \text{ V}^{-1}$ , $25,00^\circ\text{C}$ de
$\Delta G = \Delta G^\circ + RT \ln Q$	$\Delta E(\text{pil}) = E^\circ(\text{pil}) - (RT/\mathfrak{F} n) \ln Q$
$\Delta G^\circ = \Delta H^\circ - T\Delta S^\circ$	$\ln K = (n\mathfrak{F}/RT) \Delta E^\circ$
$\ln\left(\frac{K_2}{K_1}\right) = -\left(\frac{\Delta H^\circ}{R}\right)\left(\frac{1}{T_2} - \frac{1}{T_1}\right)$	$E_3^\circ = [n_1 E_1^\circ(\text{ind}) - n_2 E_2^\circ(\text{yük.})]/n_3$ $\Delta G^\circ_{\text{pil}} = -(n)(\mathfrak{F}) \Delta E^\circ_{\text{pil}}$
$K_{su} = K_a K_b$	$Q = It$

Standart İndirgenme Potansiyelleri ( $25^\circ\text{C}$  de)

Yarı tepkimeler	$E^\circ(\text{volt})$
$Au^{+}(aq) + e^- \Rightarrow Au(k)$	1.69
$MnO_4^-(aq) + 8H^+(aq) + 5e^- \rightarrow Mn^{2+}(aq) + 4H_2O(s)$	1.51
$Ag^+(aq) + 1e^- \rightarrow Ag(k)$	0.80
$Cu^{2+}(aq) + 2e^- \Rightarrow Cu(k)$	0.34
$AgCl(k) + 1e^- \rightarrow Ag(s) + Cl^-(aq)$	0.22
$Sn^{4+}(aq) + 2e^- \rightarrow Sn^{2+}(aq)$	0.15
$2H^+(aq) + 2e^- \Rightarrow H_2$	0
$Pb^{2+}(aq) + 2e^- \Rightarrow Pb(k)$	-0.13
$Sn^{2+}(aq) + 2e^- \Rightarrow Sn(k)$	-0.14
$Ni^{2+}(aq) + 2e^- \Rightarrow Ni(k)$	-0.23
$Fe^{2+}(aq) + 2e^- \rightarrow Fe(k)$	-0.44
$Cr^{3+}(aq) + 3e^- \Rightarrow Cr(k)$	-0.74
$Zn^{2+}(aq) + 2e^- \Rightarrow Zn(k)$	-0.76

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18 <sup>a</sup>	VIIIA b
IA	IIA	IIIB	IVB	VIB	VIB	VIB	VIB	VIIIB	VIIIB	IB	IB	IB	IB	VIA	VIA	VIIA	VIIA	VIIIA b
<b>The Active Metals</b>																		
1	H																	
1.008																		
3	Li	4	Be															
6.941		9.012																
11	Mg	12																
22.990		24.305																
<b>Transition Elements</b>																		
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ge	A <sub>8</sub>	Se	Br	Kr		
39.098	40.08	44.956	47.88	50.942	51.946	54.938	55.947	58.933	58.69	63.546	65.38	69.72	72.59	74.922	78.96	79.904	83.80	
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Te	I	Xe		
85.468	87.62	88.906	91.224	92.906	95.94	(98)	101.07	102.906	106.42	107.868	112.41	114.82	118.69	121.75	127.60	126.904	131.29	
55	56	57	*72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	
Cs	Ba	La	Hf	Ta	W	Re	O <sub>8</sub>	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rs	
133.905	137.33	138.905	178.49	180.948	183.85	186.21	190.2	192.22	195.08	196.966	200.59	204.38	207.2	208.98	(209)	(210)	(222)	
87	88	89	†104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	
Fr	Ra	Ac	Unq	Urp	Urn													
(223)	226.025	227.028	(261)	(262)	(263)													